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(54) Title: FORMATION OF HYDROPHILIC SITES IN PARTIALLY SILYLATED MICELLE TEMPLATED SILICA

(57) Abstract: The invention relates to a method of preparing a partially silylated silica having at a surface thereof hydrophilic sites defined by non-silylated hydroxyl groups. The method of the invention comprises the steps of (a) providing a micelle templated silica having at a surface thereof surfactant-protected hydroxyl groups and unprotected hydroxyl groups; (b) treating the micelle templated silica with a base-generating silylating agent to silylate the unprotected hydroxyl groups and thereby obtain a partially silylated micelle templated silica; and (c) treating the partially silylated micelle templated silica with an acid to displace the surfactant, thereby obtaining a partially silylated silica having at the surface thereof hydrophilic sites defined by non-silylated hydroxyl groups. Such a partially silylated silica having hydrophilic sites is useful as a catalyst support and for ion exchange in chromatography.

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## FORMATION OF HYDROPHILIC SITES IN PARTIALLY SILYLATED MICELLE TEMPLATED SILICA

### TECHNICAL FIELD

5                   The present invention pertains to improvements in the field of micelle templated silica. More particularly, the invention relates to the formation of hydrophilic sites in a partially silylated micelle templated silica.

### BACKGROUND ART

10                   The large pore of micelle templated silica in comparison to that of zeolites provides novel opportunities in the field of molecular sieves not only for the scope of treating bulkier molecules, but also for the variety of chemical modifications of their internal surface. Taking advantages of such a versatility, researchers have tried to tailor tune their acid-base, their  
15                   hydrophobicity and their catalytic properties envisionning many different applications in fields as different as adsorbents, separation and acid catalysis. In this context, the stability of micelle templated silica is an important consideration. Originally, their hydrothermal stability was poor according to the loss of their mesoporous structure in acid or alkaline  
20                   solution. Several methods have been proposed to increase the stability of mesoporous materials, including synthesis of materials with thicker pore walls, silylation, stabilization by tetralkylammonium and salt effect. New avenues explored recently the preparation of templated mesostructured materials having an organic core with an inorganic shell using  $(\text{EtO})_3\text{-Si-R-Si}(\text{OEt})_3$  type of precursors, in which R is an ethylene, phenyl or thiophene  
25                   group. However, the organic functions are located inside the walls where again accessibility restriction is expected.

### DISCLOSURE OF THE INVENTION

The present invention provides a method of forming hydrophilic sites of very small size within the channels of micelle templated silica.

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In accordance with the invention, there is thus provided a method of preparing a partially silylated silica having at a surface thereof hydrophilic sites defined by non-silylated hydroxyl groups. The method of the invention comprises the steps of:

10

a) providing a micelle templated silica having at a surface thereof surfactant-protected hydroxyl groups and unprotected hydroxyl groups;

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b) treating the micelle templated silica with a base-generating silylating agent to silylate the unprotected hydroxyl groups and thereby obtain a partially silylated micelle templated silica; and

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c) treating the partially silylated micelle templated silica with an acid to displace the surfactant, thereby obtaining a partially silylated silica having at the surface thereof hydrophilic sites defined by non-silylated hydroxyl groups.

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The present invention also provides, in another aspect thereof, a partially silylated silica having at a surface thereof hydrophilic sites defined by non-silylated hydroxyl groups.

As used herein, the expression "base-generating silylating agent" refers to a silylating agent which is capable of forming a base as a by-product of the silylation. Applicant has found quite unexpectedly that such a

base does not displace the templating surfactant so that the surfactant-protected hydroxyl groups remain protected during the silylation and a partial silylation of the micelle templated silica can thus be achieved. In contrast, the acid formed as a by-product during silylation by acid-generating silylating agents such as, for example, chlorotrimethylsilane displaces the templating surfactant, leading to a complete silylation and to a very hydrophobic surface. Once the micelle templated silica has been partially silylated, it can thereafter be treated with an acid to displace the templating surfactant and thereby obtain the desired partially silylated silica having hydrophilic sites at the surface thereof.

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#### MODES FOR CARRYING OUT THE INVENTION

Examples of suitable base-generating silylating agents which may be used for effecting the partial silylation include hexamethyldisilazane, di-n-butyltetramethyldisilazane, 1,3-divinyl-1,3-diphenyl-1,3-dimethyldisilazane, hexamethyldisiloxane, 1,3-diallyltetramethyldisiloxane, 1,3-divinyl-1,3-diphenyl-1,3-dimethyldisiloxane, triphenylsilanol, diphenylsilanediol, bis(cyanopropyl)tetramethyldisiloxane, N,O-bis(trimethylsilyl)acetamide and N,O-bis(trimethylsilyl)trifluoroacetamide. Hexamethyldisilazane is particularly preferred.

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According to a preferred embodiment of the invention, step (b) is carried out by treating the micelle templated silica under reflux at a temperature of about 25° - 150°C in a solution of the silylating agent in a non-polar solvent. The solvent used for dissolving the silylating agent must be non-polar in order to prevent a dissolution of the templating surfactant. Examples of suitable non-polar solvents which may be used include

toluene, benzene, cyclohexane, n-hexane, trichloromethane and diethylether. Toluene is particularly preferred.

According to another preferred embodiment, step (b) is carried  
5 out by treating the micelle templated silica in a fluidized bed under a flow of a inert gas saturated at about 50° - 150°C with the silylating agent. Nitrogen is preferably used as inert gas.

According to a further preferred embodiment, step (c) is  
10 carried out by washing the partially silylated micelle templated surfactant with an acid in admixture with a polar solvent such as ethanol. Hydrochloric acid is preferably used.

The surfactant used for protecting the hydroxyl groups at the  
15 surface of the silica is preferably a quaternary ammonium salt. Examples of suitable quaternary ammonium salts which may be used include tetramethylammonium salts, cetyltrimethylammonium salts and benzyltrimethylammonium salts. Cetyltrimethylammonium bromide is particularly preferred. It is also possible to use a quaternary phosphonium  
20 salt such as, for example, dodecyltriphenylphosphonium bromide.

The partially silylated silica having hydrophilic sites and obtained by the method according to the invention is useful as a catalyst support and for ion exchange in chromatography. Typically, the hydrophilic  
25 sites represent about 35% to about 55% of the surface of the silica. The hydroxyl groups of the partially silylated silica are preferably silylated by trimethylsilyl groups. Silylation and particularly trimethylsilylation enhance



the mechanical stability of the silica. The hydrophilic sites, on the other hand, are available for further surface modifications.

The following non-limiting examples illustrate the invention.

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#### Preparation of Micelle Templated Silica

Pure hexagonal micelle templated silica was prepared from a gel of molar composition: 1.00 SiO<sub>2</sub>, 0.86 Na<sub>2</sub>O, 0.44 (TMA)<sub>2</sub>O, 0.30 CTMABr, 63.3 H<sub>2</sub>O (TMA = tetramethylammonium; CTMABr = cetyltrimethylammonium bromide). A solution of cetyltrimethylammonium bromide was slowly added to a clear gel containing fumed silica (Cab-O-Sil), sodium silicate and TMA-silicate with vigorous stirring at room temperature. The resulting gel was transferred into a Teflon-lined autoclave (1 litre autoclave for about 60 g of silica) and maintained for 24 h at 130°C. The resulting powder was filtered, washed with distilled water and dried in air. To improve the long range order of the as-synthesized material, the resulting solid was treated in 600 ml of water per 60 g of solid into a Teflon-lined autoclave for 24 h at 130°C. Then, this powder was filtered, washed with distilled water, and dried in air.

#### EXAMPLE 1

The micelle templated silica (1.0 g) as prepared above was treated under reflux in a solution of hexamethyldisilazane in toluene and allowed to react for 2 h at 110°C. The silylated product was washed with ethanol and dried in air. The solid (250 mg) was added to a mixture of 100 ml of ethanol and 10 ml of 0.1 N HCl and stirred for two hours. Under

these conditions, the cetyltrimethylammonium groups were removed from the solid. The acid washed material was filtered and washed with ethanol. The excess of HCl was titrated with 0.1 N NaOH.

5    EXAMPLE 2

The micelle templated silica (1.2 g) as prepared above was treated in a fluidized bed under a gas flow (15 cm<sup>3</sup>/min.) of nitrogen saturated at 130°C with hexamethyldisilazane; 5 ml of the latter were consumed. The silylated product was washed with ethanol and dried in air.

10   The solid (250 mg) was added to a mixture of 100 ml of ethanol and 10 ml of 0.1 N HCl and stirred for two hours. Under these conditions, the cetyltrimethylammonium groups were removed from the solid. The acid washed material was filtered and washed with ethanol. The excess of HCl was titrated with 0.1 N NaOH.

15

EXAMPLE 3 (Comparative)

The micelle templated silica (1.0 g) was treated overnight under reflux at 100°C in 20 ml of a 1:1 mixture of chlorotrimethylsilane and hexamethyldisiloxane. The silylated product was washed with ethanol and dried in air. The solid (250 mg) was added to a mixture of 100 ml of ethanol and 10 ml of 0.1 N HCl and stirred for two hours. Under these conditions, the cetyltrimethylammonium groups were removed from the solid. The acid washed material was filtered and washed with ethanol. The excess of HCl was titrated with 0.1 N NaOH.

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The silylated and acid washed materials obtained in Examples 1, 2 and 3 were all tested for their ion exchange capacity. A chloride salt of

the  $(\text{Co(en)}_2\text{Cl}_2)^+$  complex was cation exchanged at room temperature for one hour. Typically, for 200 mg of the silylated and acid washed product, 3 ml of concentrated ammonia (30%) were added to a 50 ml solution of cobalt complex ( $2.8 \times 10^{-3}$  M) in a 73:27 water:ethanol mixture. Except for the fully silylated silica obtained in Example 3, which remained white, the partially silylated solids obtained in Examples 1 and 2 took up the coloration of the cobalt complex during ion exchange at pH 10.

The  $\text{SiO}^-/\text{Si}$  ratio was also determined by acid-base titration in a non-aqueous solvent such as ethanol. The  $\text{SiO}^-/\text{Si}$  ratio decreases from 17.8 to 13.8, 9.4 and 0 for the micelle templated silica to the silylated forms obtained in Examples 1, 2 and 3, respectively.

The tethered trimethylsilane groups are characterized by characteristic IR bands at 1250, 850, 750  $\text{cm}^{-1}$ . The deepness of silylation measured by  $\text{SiO}^-$  titrations was quantitatively confirmed by  $^{13}\text{C}$  and  $^{29}\text{Si}$  MAS-NMR spectroscopy.

The silylation deepness calculated in terms of the number of trimethylsilyl groups per  $\text{nm}^2$  was obtained from the elemental analysis of carbon performed on the silylated and acid washed materials of Examples 1, 2 and 3, according to the following equation:

$$S = \left[ \frac{(\%C)}{50 - (\%C)} \right] \frac{N_A}{(TRMS)(S_{MTS})} \times 10^{-18}$$



where %C is the content of carbon reported in weight percentage,  $N_A$  is Avogadro's number, TRMS is the effective molecular weight of the trimethylsilyl groups and  $S_{MTS}$  is the surface area of non-silylated material (1060 m<sup>2</sup>/g in the present experiment). The results are reported in the following Table.

TABLE

Solid Tested	S TRMS/nm <sup>2</sup>	TRMS coverage (%)
Ex. 1	1.1	42
Ex. 2	1.6	61
Ex. 3	2.6	100

## CLAIMS

1. A method of preparing a partially silylated silica having at a surface thereof hydrophilic sites defined by non-silylated hydroxyl groups, comprising the steps of:

a) providing a micelle templated silica having at a surface thereof surfactant-protected hydroxyl groups and unprotected hydroxyl groups;

b) treating said micelle templated silica with a base-generating silylating agent to silylate the unprotected hydroxyl groups and thereby obtain a partially silylated micelle templated silica; and

c) treating said partially silylated micelle templated silica with an acid to displace the surfactant, thereby obtaining a partially silylated silica having at the surface thereof hydrophilic sites defined by non-silylated hydroxyl groups.

2. A method according to claim 1, wherein said silylating agent is selected from the group consisting of hexamethyldisilazane, di-n-butyl-tetramethyldisilazane, 1,3-divinyl-1,3-diphenyl-1,3-dimethyldisilazane, hexamethyldisiloxane, 1,3-diallyltetramethyldisiloxane, 1,3-divinyl-1,3-diphenyl-1,3-dimethyldisiloxane, triphenylsilanol, diphenylsilanediol, bis(cyanopropyl)tetramethyldisiloxane, N,O-bis(trimethylsilyl)acetamide and N,O-bis(trimethylsilyl)trifluoroacetamide.

3. A method according to claim 2, wherein said silylating agent is hexamethyldisilazane.

4. A method according to any one of claims 1 to 3, wherein step (b) is carried out by treating said micelle templated silica under reflux at a tempera-

ture of about 25° - 150°C in a solution of said silylating agent in a non-polar solvent.

5. A method according to claim 4, wherein said non-polar solvent is selected from the group consisting of toluene, benzene, cyclohexane, n-hexane, trichloromethane and diethylether.

6. A method according to claim 5, wherein said non-polar solvent is toluene.

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~~7. A method according to any one of claims 1 to 3, wherein step (b) is carried out by treating said micelle templated silica in a fluidized bed under a flow of an inert gas saturated at about 50° - 150°C with said silylating agent.~~

8. A method according to claim 7, wherein said inert gas is nitrogen.

9. A method according to any one of claims 1 to 8, wherein step (c) is carried out by washing said partially silylated micelle templated silica with said acid in admixture with a polar solvent.

10. A method according to claim 9, wherein said acid is hydrochloric acid and said polar solvent is ethanol.

11. A method according to any one of claims 1 to 10, wherein the surfactant-protected hydroxyl groups are protected by quaternary ammonium groups.

12. A method according to claim 11, wherein said quaternary ammonium groups are tetramethylammonium, cetyltrimethylammonium or benzyltrimethylammonium groups.

13. A method according to claim 12, wherein said quaternary ammonium groups are cetyltrimethylammonium groups.

14. A method according to any one of claims 1 to 10, wherein the surfactant-protected hydroxyl groups are protected by quaternary phosphonium groups.

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15. A method according to claim 14, wherein said quaternary phosphonium are dodecyltriphenylphosphonium groups.

16. A partially silylated silica having at a surface thereof hydrophilic sites defined by non-silylated hydroxyl groups.

17. A silica according to claim 16, wherein the hydrophilic sites represent about 35% to about 55% of said surface.

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18. A silica according to claim 16 or 17, wherein hydroxyl groups are silylated by trimethylsilyl groups.

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/CA 01/01161

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C01B37/02 C01B33/12 C09C1/30

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C01B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the International search (name of data base and, where practical, search terms used)

WPI Data, PAJ, INSPEC, COMPENDEX, EPO-Internal, CHEM ABS Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	BARTHEL H ET AL: "Determination of trimethylchlorosilane modified silica surface properties by adsorption of hexamethyldisiloxane" COMPOSITE INTERFACES, vol. 6, no. 1, 1999, pages 27-34, XP001051211 Zeist, Netherlands page 28, "Experimental"	16-18
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☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents :

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
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- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
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**C.(Continu**

Category \*

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Relevant to claim No.

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claims 1-10

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# INTERNATIONAL SEARCH REPORT

Information on patent family members

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(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

(71) Applicant (*for all designated States except US*): UNIVERSITE LAVAL [CA/CA]; Cité Universitaire, Québec, Québec G1K 7P4 (CA).

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(74) Agents: SWABEY OGILVY RENAULT et al.; 1981 McGill College Avenue, Suite 1600, Montreal, Québec H3A 2Y3 (CA).

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(57) Abstract: The invention relates to a method of preparing a partially silylated silica having at a surface thereof hydrophilic sites defined by non-silylated hydroxyl groups. The method of the invention comprises the steps of (a) providing a micelle templated silica having at a surface thereof surfactant-protected hydroxyl groups and unprotected hydroxyl groups; (b) treating the micelle templated silica with a base-generating silylating agent to silylate the unprotected hydroxyl groups and thereby obtain a partially silylated micelle templated silica; and (c) treating the partially silylated micelle templated silica with an acid to displace the surfactant, thereby obtaining a partially silylated silica having at the surface thereof hydrophilic sites defined by non-silylated hydroxyl groups. Such a partially silylated silica having hydrophilic sites is useful as a catalyst support and for ion exchange in chromatography.

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## AMENDED CLAIMS

[received by the International Bureau on 3 April 2002 (03.04.02);  
original claim 16 amended; remaining claims unchanged (1 page )]

12. A method according to claim 11, wherein said quaternary ammonium groups are tetramethylammonium, cetyltrimethylammonium or benzyltrimethylammonium groups.

13. A method according to claim 12, wherein said quaternary ammonium groups are cetyltrimethylammonium groups.

14. A method according to any one of claims 1 to 10, wherein the surfactant-protected hydroxyl groups are protected by quaternary phosphonium groups.

15. A method according to claim 14, wherein said quaternary phosphonium are dodecyltriphenylphosphonium groups.

16. A partially silylated silica obtained by a method as defined in any one of claims 1 to 15, said partially silylated silica having at a surface thereof uniformly distributed hydrophilic sites defined by non-silylated hydroxyl groups.

17. A silica according to claim 16, wherein the hydrophilic sites represent about 35% to about 55% of said surface.

18. A silica according to claim 16 or 17, wherein hydroxyl groups are silylated by trimethylsilyl groups.

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